

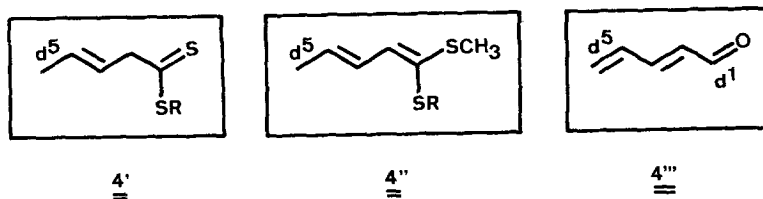
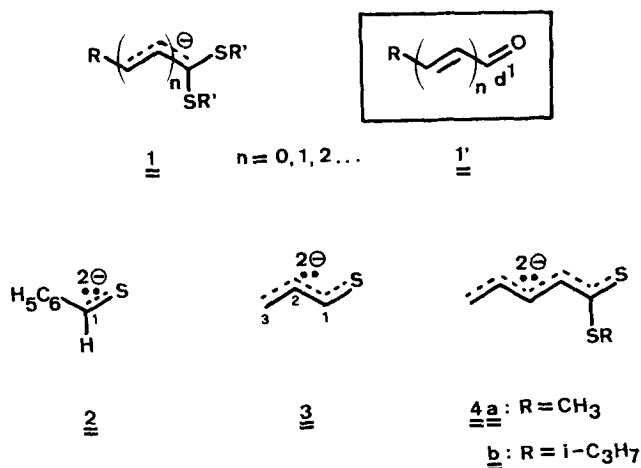
DIANION DERIVATIVES OF METHYL- AND ISOPROPYL-2.4-PENTADIENEDITHIOATE AS d^5 -REAGENTS

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Summary: Li-K-Derivatives of the dianions 4 are generated. Reactions with carbonyl compounds occur at the ω -position to yield 6 - 9.

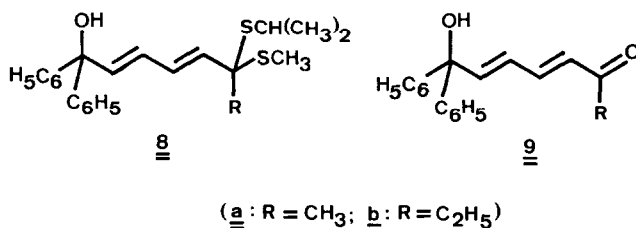
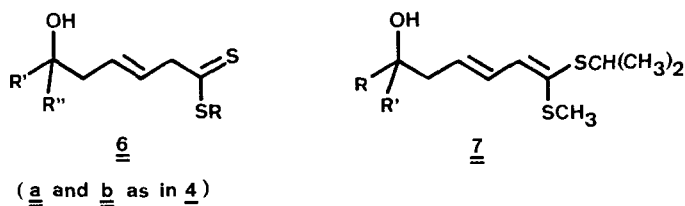
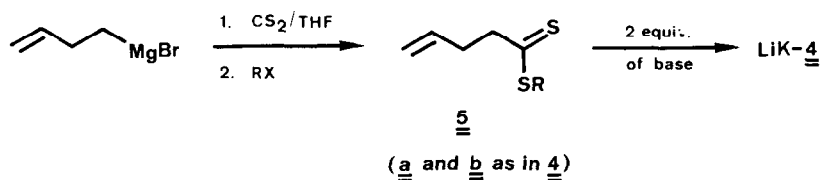
Deprotonated thioacetals 1 show a pronounced preference to react at the thiolated carbon atom with electrophiles¹⁾. Thus, they are widely used in transformations requiring d^1 -synthons^{1,2)} 1'. In many synthetic situations, reactivity remote from the functional group is desired. We



have previously shown³⁾ that this can be accomplished by employing dianion derivatives which, in addition to exhibiting ω -ambidoselectivity²⁾, have reactivity umpolung as compared with their neutral counterparts (π -systems vs. doubly "LUMO-filled" π -systems)²⁾. Two examples involving sulfur are Li₂-thiobenzaldehyde^{3a,3c)} 2 and Li₂-thioacrolein^{3b,3c)} 3. The present

communication demonstrates that the Li/K-dithioate 4 corresponds to the d^5 -synthons²⁾ 4', 4'', and 4'''.

Addition of carbon disulfide to homoallyl Grignard reagent and alkylation of the resulting dithioate with iodomethane or 2-iodopropane (in the presence of HMPT) furnishes the dithioesters 5 in better than 80% yield⁴⁾. As with γ,δ -unsaturated ketones^{3d)}, treatment of 5 first with potassium hydride (0° - 20°C) and then with *sec*-butyllithium/2 TMEDA (-78°C) in THF generates Li/K-derivatives of dianions: 4a as orange suspension, 4b as red-brown solution. Both



combine with aldehydes and ketones preferentially⁵⁾ in the 5-position to give dithioates of type 6 after aqueous workup or ketene thioacetals⁶⁾ 7 after quenching with iodomethane; the better yields are obtained with 4b.

Deprotonation (2 LDA, THF/15% HMPT, -78° - 0°C) and alkylation of the benzophenone adduct 7, R = R' = C₆H₅, leads to the 6-hydroxy-dienone thioacetals 8 (cf. ref.^{1b)}) which were hydrolyzed (CH₃I/aq. THF-acetonitrile, CaCO₃)^{1a)} to the parent carbonyl compounds 9, cf. 4'''.

The yields and structures of chromatographed (SiO₂) and analytically pure products are listed in the accompanying table, together with some characteristic data.

Table

Yields and some data of the products 5 - 9.

precursors	products	yield [%]	b.p., m.p. or $^1\text{H-NMR}$ (δ in ppm)
1-bromo-4-butene	<u>5a</u>	89	43-45 ⁰ /1 Torr
1-bromo-4-butene	<u>5b</u>	81	38-40 ⁰ /0.6 Torr
<u>5a</u> , benzaldehyde	<u>6a</u> , R' = C ₆ H ₅ , R'' = H	43	CHO, δ = 4.58 (t, J = 7 Hz)
<u>5a</u> , cyclohexanone	<u>6a</u> , R' - R'' = (CH ₂) ₅	38	-(CH ₂) ₅ -, δ = 1.4 (m)
<u>5a</u> , benzophenone	<u>6a</u> , R' = R'' = C ₆ H ₅	71	64-66 ⁰ C
<u>5b</u> , propanal	<u>6b</u> , R' = C ₂ H ₅ , R'' = H	56	CHO, δ = 3.9 (quint, J=7Hz)
<u>5b</u> , 2-methylpropanal	<u>6b</u> , R' = i-C ₃ H ₇ , R'' = H	49	CHO, δ = 3.28 (m)
<u>5b</u> , 2.2-dimethylpropanal	<u>6b</u> , R' = t-C ₄ H ₉ , R'' = H	60	(CH ₃) ₃ C, δ = 0.9 (s)
<u>5b</u> , benzaldehyde	<u>6b</u> , R' = C ₆ H ₅ , R'' = H	73	CHO, δ = 4.62 (t, J = 7 Hz)
<u>5b</u> , acetone	<u>6b</u> , R' = R'' = CH ₃	39	(CH ₃) ₂ C, δ = 1.16 (s)
<u>5b</u> , cyclopentanone	<u>6b</u> , R' - R'' = (CH ₂) ₄	44	-(CH ₂) ₄ -, δ = 1.6 (m)
<u>5b</u> , cyclohexanone	<u>6b</u> , R' - R'' = (CH ₂) ₅	55	-(CH ₂) ₅ -, δ = 1.43 (m)
<u>5b</u> , benzophenone	<u>6b</u> , R' = R'' = C ₆ H ₅	84	53-55 ⁰ C
<u>5b</u> , benzophenone, CH ₃ I	<u>7</u> , R = R' = C ₆ H ₅	80	85-86 ⁰ C
<u>7</u> , R = R' = C ₆ H ₅ , CH ₃ I	<u>8a</u>	76	CH ₃ C, δ = 1.7 (s)
<u>7</u> , R = R' = C ₆ H ₅ , C ₂ H ₅ I	<u>8b</u>	85	CH ₃ CH ₂ , δ =1.0 (t, J = 7 Hz)
<u>8a</u> , alkylative hydrolysis	<u>9a</u>	57	128-130.5 ⁰ C
<u>8b</u> , alkylative hydrolysis	<u>9b</u>	39	110-112 ⁰ C

REFERENCES AND FOOTNOTES

- 1) a) Reviews: D. Seebach, *Angew. Chem.* 81, 690 (1969); *ibid.* *Int. Ed. Engl.* 8, 639 (1969); D. Seebach, *Synthesis*, 17 (1969); D. Seebach and K.-H. Geiss, *J. Organomet. Chemistry, Library 1*, Elsevier Scientific Publishing Company, Amsterdam, 1976, p. 1 - 92; D. Seebach, K.-H. Geiss, M. Kolb and A.K. Beck, *Modern Synthetic Methods*, 1976, p. 173 - 299, Schweiz. Chemiker-Verband Zürich; B.-T. Gröbel and D. Seebach, *Synthesis*, 357 (1977);
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- 3) a) D. Seebach and K.-H. Geiss, *Angew. Chem.* 86, 202 (1974); *ibid.* *Int. Ed. Engl.* 13, 202 (1974);
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- 4) J. Meijer, P. Vermeer and L. Brandsma, *Rec. Trav. Chim.* 92, 601 (1973).
- 5) A few percents of d^3 -product are also formed, but readily removed during chromatographic purification of 6.
- 6) Only one stereoisomer of 7 is isolated as judged from the $^1\text{H-NMR-SCH}_3$ singlett. In analogy with the configurations of the products 3b,3c from 3, we have drawn E,E-7.

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